## Samarium(II) Diiodide Induced Polarity Inversion of $\pi$ -Allyl Palladium Complexes: The Formation of Allylic Selenides

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Abstract: Allylic acetates were reduced with Pd(0)-SmI<sub>2</sub> in the presence of ArSeBr to form corresponding allylic selenides in good yields.

**Keywords:** Allyl acetates, allylic selenides, arylselenenyl bromides, palladium complexes, polarity inversion.

As a versatile one-electron transfer reducing and coupling agent, samarium diiodide has been extensively used in organic synthesis in last decade. It has been reported that electrophilic  $\pi$ -allyl palladium complexes could undergo polarity inversion to nucleophilic species with the aid of SmI<sub>2</sub>, and could trap some electrophiles<sup>1</sup>. It is well known that allylic selenides are rather important intermediates in organic synthesis. They have been recognized as useful synthons of allylic anion stabilized by seleno group, and they are regioselectively  $\alpha$ -alkylated<sup>2,3</sup>. In literature a variety of synthetic methods for allylic selenides have been reported such as nucleophilic substitution of allylic halides with phenylseleno anion<sup>4,5</sup>. However, those methods have the disadvantages of being in low yields or long reaction time.

We found that the SmI<sub>2</sub> induced polarity inversion of  $\pi$ -allyl palladium complex could be an efficient method for preparation of allylic selenides from electrophilic arylselenenyl bromides and allylic acetates (Scheme 1). In this reaction, we added arylselenenyl bromides 2 to the mixed solution of SmI<sub>2</sub>, Pd(PPh<sub>3</sub>)<sub>4</sub> and allylic acetates 1 in THF, corresponding allylic selenides 3 were formed after being stirred for 4 h at room temperature in good yields. The results are listed in **Table 1**. The structure of compounds **3a-3e** was confirmed by IR and <sup>1</sup>H NMR<sup>6</sup>. The mechanism of reaction was proposed as follows: Allyl acetate reacted with Pd(PPh<sub>3</sub>)<sub>4</sub> to form  $\pi$ -allyl palladium complex. The complex was converted to allyl anion *via* reduction with SmI<sub>2</sub>. Then the allyl anion trapped ArSeBr to yield allylic selenides.

General procedure: A solution of  $SmI_2$  (5 mmol) in THF (30 mL) was added to allyl acetate **1** (2 mmol) in anhydrous THF (3 mL), followed by Pd(PPh<sub>3</sub>)<sub>4</sub> (0.05 mmol) under

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 $R \longrightarrow OAc + ArSeBr \xrightarrow{Pd(PPh_3)_4 - SmI_2} R \xrightarrow{SeAr}$ 

the nitrogen atmosphere. The mixture was stirred at room temperature for 0.5 h. Then phenylselenenyl bromide 2 (2.2 mmol) in THF (10 mL) was added dropwise to it. The resulting mixture was continued to be stirred at room temperature for 4 h (the reaction was mornitored by TLC). After evaporation of THF, the crude product was diluted with ethyl ether (30 mL), washed with saturated solution of NaCl (3×30 mL), dried over Na<sub>2</sub>SO<sub>4</sub> and concentrated. The crude product was purified by TLC on silica gel to form corresponding allylic selenide 3 in 58% yield.

 Table 1
 SmI2 induced synthesis of allylic selenides catalyzed by palladium complexes

Entry	R	Ar	Time (h)	Isolated yield <sup>a</sup> (%)	mp (°C)	Lit. mp (°C)
3a	Н	$C_6H_5$	4.5	58	oil	oil
3b	Н	$4\text{-}CH_3C_6H_4$	4.5	53	oil	oil
3c	Н	$4-CH_3OC_6H_4$	4.5	51	oil	oil
3d	$\mathrm{C}_{6}\mathrm{H}_{5}$	$C_6H_5$	5	49	62-63	64-65 <sup>7</sup>
3e	$\mathrm{C}_{6}\mathrm{H}_{5}$	$4\text{-}CH_3C_6H_4$	5	47	53-54	54.5-55.5 <sup>7</sup>

In summary, a new route for prepare allylic selenides has been provided utilizing samarium diiodide induced polarity inversion of  $\pi$ -allyl palladium complexes under mild and neutral condition.

## **References and Notes**

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- 6. <sup>1</sup>H NMR (CDCl<sub>3</sub>,  $\delta_{\rm H}$  ppm): **3a**: 3.20-3.33 (d, 2H, *J*=7Hz), 4.63-4.80 (m, 2H), 5.43-6.10 (m, 1H), 6.93-7.33 (m, 5H). **3b**: 2.17 (s, 3H), 3.17-3.30 (d, 2H, *J*=7Hz), 4.56-4.78 (m, 2H), 5.23-6.03 (m, 1H), 6.70-7.18 (m, 4H). **3c**: 3.63 (s, 3H), 3.17-3.28 (d, 2H, *J*=6Hz), 4.61-4.76 (m, 2H), 5.40-5.96 (m, 1H), 6.47-7.30 (m, 4H). **3d**: 3.43-3.53 (m, 2H), 5.83-6.12 (m, 2H), 7.13-7.33 (m, 5H). **3e**: 2.23 (s, 3H), 3.43-3.57 (m, 2H), 5.97-6.23 (m, 2H), 6.73-7.37 (m, 4H).

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